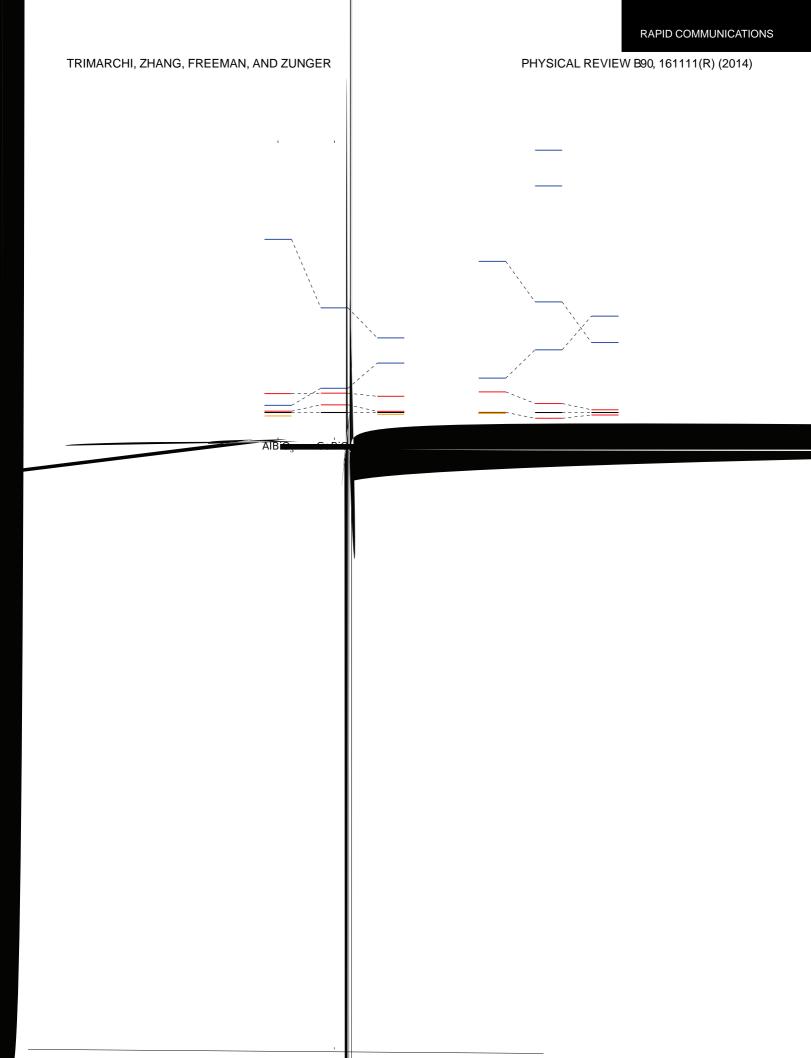
= A , Ga, In, Sc, Y, and La), \quad e c $\,$ b c A

3 . . . ,



STRUCTURALLY UNSTABLE A^{III} BiO₃

Perovskite structures where the octahedra are allowed to tilt: relative stability of the two Pnma GdFeQype perovskite variants. A whole range of distorted perovskite structure types can be obtained starting from the ideal cubic perovskite by octahedra rotations and cation shifts. These distorted structures can lower the energy by optimizing the atom packing. The Pnma GdFeQ structure-type in particular is the perovskite in which the octahedra can tilt around a generic axis and distort from the ideal regular shape. As shown in Fig. 1, the energy of the A^{III} BiO_3 compounds decreases considerably going from the m3m to the Pnma perovskite structure, regardless of the Bi position. The con guration with Bi at the cavity site is still energetically preferred, but with an energy distance from the con guration with Bi at the site much reduced with respect to them3m perovskite and as low as 0.1 eV/at. Moreover, the GdFeQype structure with the A^{III} cation at the octahedral site, is close in energy to the experimentally observed structure of AIBiand InBia and degenerate with this in the case of ScBiO

Electronic structure of the GdFeQype perovskitesThe A^{III} BiO $_3$ systems in thePnma perovksite structure are insulators but without the-p band inversion observed in the Pm3m form. Figure 3 compares the GGASOC band structure of InBiQ in the Pm3m and Pnma perovskite structures with Bi at the octahedral site. While InBiO the Pm3m structure is a metal with the Bi-p band inversion at theR point at about 3 eV above the Fermi level, the octahedral tilting and internal distortions that produce the GdFatOpe structure open a band gap but remove the inversion. We nd that all A^{III} BiO $_3$ systems in thisPnma perovskite are trivial insulators withZ $_2$ = 0 irrespective of the Bi position.

GSGO-predicted and experimental structures.

STRUCTURALLY UNSTABLE A $^{\mbox{\scriptsize III}}$ BiO $_3$...

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position, metallic or trivial insulators in the cubic perovskite structure which is, however, unstable. When the octahedra are allowed to tilt, the cubic perovskite relaxes into distorted perovskite phases of much lower energy. However, we nd